

# Periodic Table of Elements

## Selected Phase Transition Temperatures

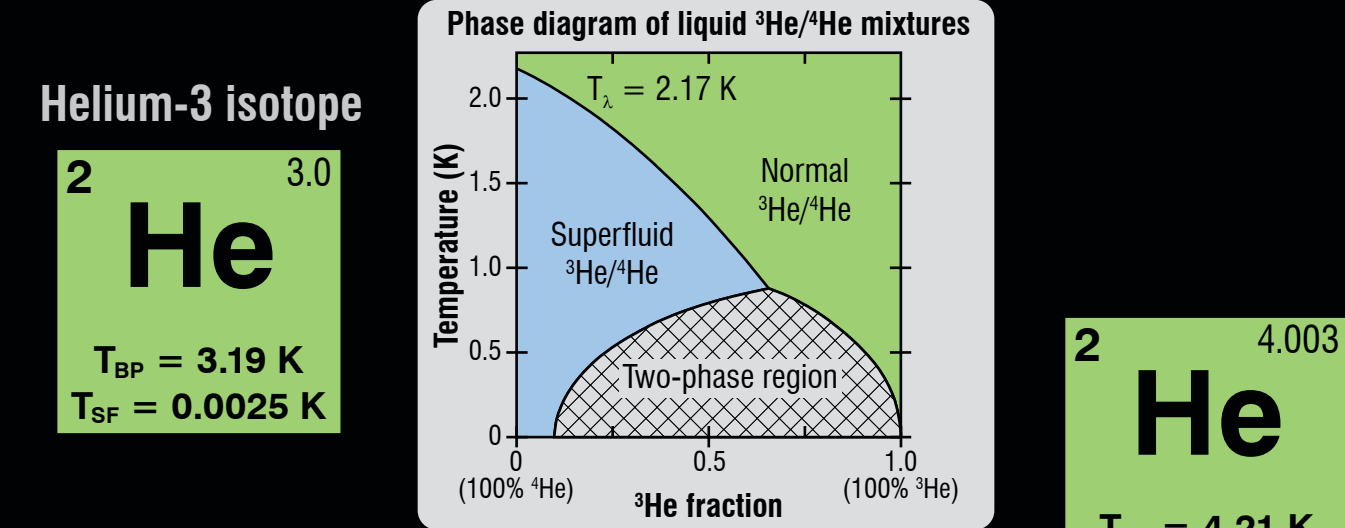
1 1.008 <b>H</b> $T_{BP} = 20.3 \text{ K}$ $T_{MP} = 13.8 \text{ K}$	2 9.012 <b>He</b> $T_{BP} = 4.21 \text{ K}$ $T_{\lambda} = 2.17 \text{ K}$																
3 6.941 <b>Li</b> $T_S = 0.0004 \text{ K}$	4 9.012 <b>Be</b> $T_S = 0.026 \text{ K}$																
11 22.99 <b>Na</b> Sodium	12 24.31 <b>Mg</b> Magnesium																
19 39.10 <b>K</b> Potassium	20 40.08 <b>Ca</b> Calcium																
21 44.96 <b>Sc</b> Scandium	22 47.87 <b>Ti</b> $T_S = 0.39 \text{ K}$	23 50.94 <b>V</b> $T_S = 5.38 \text{ K}$	24 52.00 <b>Cr</b> $T_N = 311 \text{ K}$	25 54.94 <b>Mn</b> $(\alpha) T_N = 100 \text{ K}$	26 55.85 <b>Fe</b> $T_C = 1043 \text{ K}$	27 58.93 <b>Co</b> $T_C = 1388 \text{ K}$	28 58.69 <b>Ni</b> $T_C = 627 \text{ K}$	29 63.55 <b>Cu</b> Copper	30 65.41 <b>Zn</b> $T_S = 0.875 \text{ K}$	31 69.72 <b>Ga</b> $T_S = 1.091 \text{ K}$	32 72.64 <b>Ge</b> Germanium	33 74.92 <b>As</b> Arsenic	34 78.96 <b>Se</b> Selenium	35 79.90 <b>Br</b> Bromine	36 83.80 <b>Kr</b> $T_{BP} = 119.9 \text{ K}$ $T_{MP} = 115.8 \text{ K}$		
37 85.47 <b>Rb</b> Rubidium	38 87.62 <b>Sr</b> Strontium	39 88.91 <b>Y</b> Yttrium	40 91.22 <b>Zr</b> $T_S = 0.546 \text{ K}$	41 92.91 <b>Nb</b> $T_S = 9.20 \text{ K}$	42 95.94 <b>Mo</b> $T_S = 0.92 \text{ K}$	43 98 <b>Tc</b> $T_S = 7.77 \text{ K}$	44 101.1 <b>Ru</b> $T_S = 0.51 \text{ K}$	45 102.9 <b>Rh</b> $T_S = 0.00033 \text{ K}$	46 106.4 <b>Pd</b> Palladium	47 107.9 <b>Ag</b> Silver	48 112.4 <b>Cd</b> $T_S = 0.52 \text{ K}$	49 114.8 <b>In</b> $T_S = 3.404 \text{ K}$	50 118.7 <b>Sn</b> $T_S = 3.722 \text{ K}$	51 121.8 <b>Sb</b> Antimony	52 127.6 <b>Te</b> Tellurium	53 126.9 <b>I</b> Iodine	54 131.3 <b>Xe</b> $T_{BP} = 165.0 \text{ K}$ $T_{MP} = 161.3 \text{ K}$
55 132.9 <b>Cs</b> Cesium	56 137.3 <b>Ba</b> Barium		72 178.5 <b>Hf</b> $T_S = 0.12 \text{ K}$	73 180.9 <b>Ta</b> $T_S = 4.483 \text{ K}$	74 183.8 <b>W</b> $T_S = 0.012 \text{ K}$	75 186.2 <b>Re</b> $T_S = 1.4 \text{ K}$	76 190.2 <b>Os</b> $T_S = 0.655 \text{ K}$	77 192.2 <b>Ir</b> $T_S = 0.14 \text{ K}$	78 195.1 <b>Pt</b> Platinum	79 197.0 <b>Au</b> Gold	80 200.6 <b>Hg</b> $(\alpha) T_S = 4.153 \text{ K}$	81 204.4 <b>Tl</b> $T_S = 2.39 \text{ K}$	82 207.2 <b>Pb</b> $T_S = 7.193 \text{ K}$	83 209.0 <b>Bi</b> Bismuth	84 (209) <b>Po</b> Polonium	85 (210) <b>At</b> Astatine	86 (222) <b>Rn</b> $T_{BP} = 211.5 \text{ K}$ $T_{MP} = 202.2 \text{ K}$
87 (223) <b>Fr</b> Francium	88 (226) <b>Ra</b> Radium		104 (261) <b>Rf</b> Rutherfordium	105 (262) <b>Db</b> Dubnium	106 (266) <b>Sg</b> Seaborgium	107 (264) <b>Bh</b> Bohrium	108 (277) <b>Hs</b> Hassium										

Atomic number → 26 ← Atomic weight  
Element symbol → **Fe** ← Color code  
 $T_C = 1043 \text{ K}$  ← Transition temperature

Superconductivity:  $T_S =$  Critical temperature  
Ferromagnetism:  $T_C =$  Curie temperature  
Antiferromagnetism:  $T_N =$  Néel temperature  
No known ordering

Gaseous elements phase change at 1 bar pressure  
 $T_{\lambda}$  = Helium-4 superfluidity  
 $T_{SF}$  = Helium-3 superfluidity  
 $T_{BP}$  = Boiling point  
 $T_{MP}$  = Melting point

273.15 K = 0 °C    1 GPa = 10 kbar



**Magnetic standard reference materials**

Platinum  $\chi_{DC}(297 \text{ K}) = 1.01 \times 10^{-6} \text{ emu/g-Oe}$  (NIST SRM #764a)  
 Palladium  $\chi_{DC}(298 \text{ K}) = 5.25 \times 10^{-6} \text{ emu/g-Oe}$  (Quantum Design part #4500-658)  
 Nickel  $M_{sat}(298 \text{ K}, 5000 \text{ Oe}) = 54.9 \text{ emu/g}$  (NIST SRM #772a)

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 Dedicated to the memory of Bernd Matthias

**References:**

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57 138.9 <b>La</b> $(fcc) T_S = 6.00 \text{ K}$	58 140.1 <b>Ce</b> $T_{N-hex} = 13.7 \text{ K}$	59 140.9 <b>Pr</b> $T_{N-hex} = 0.03 \text{ K}$	60 144.2 <b>Nd</b> $T_{N-hex} = 19.9 \text{ K}$ $T_{N-cub} = 7.5 \text{ K}$	61 (145) <b>Pm</b> Promethium	62 150.4 <b>Sm</b> $T_{N-hex} = 109 \text{ K}$ $T_{N-cub} = 14.0 \text{ K}$	63 152.0 <b>Eu</b> $T_{N-cub} = 90.4 \text{ K}$	64 157.3 <b>Gd</b> $T_C = 293 \text{ K}$	65 158.9 <b>Tb</b> $T_{N-hex} = 230 \text{ K}$ $T_C = 220 \text{ K}$	66 162.5 <b>Dy</b> $T_{N-hex} = 180 \text{ K}$ $T_C = 90.5 \text{ K}$	67 164.9 <b>Ho</b> $T_{N-hex} = 132 \text{ K}$ $T_C = 19.5 \text{ K}$	68 167.3 <b>Er</b> $T_{N-hex} = 85 \text{ K}$ $T_C = 18.7 \text{ K}$	69 168.9 <b>Tm</b> $T_{N-hex} = 58 \text{ K}$ $T_C = 32 \text{ K}$	70 173.0 <b>Yb</b> Ytterbium	71 175.0 <b>Lu</b> Lutetium
89 (227) <b>Ac</b> Actinium	90 232.0 <b>Th</b> $T_S = 1.368 \text{ K}$	91 231.0 <b>Pa</b> $T_S = 1.4 \text{ K}$	92 238.0 <b>U</b> $(\beta) T_S = 0.8 \text{ K}$	93 (237) <b>Np</b> Neptunium	94 (244) <b>Pu</b> Plutonium	95 (243) <b>Am</b> $T_S = 0.79 \text{ K}$	96 (247) <b>Cm</b> $T_N = 64 \text{ K}$	97 (247) <b>Bk</b> $T_N = 34 \text{ K}$	98 (251) <b>Cf</b> $T_C = 51 \text{ K}$	99 (252) <b>Es</b> Einsteinium	100 (257) <b>Fm</b> Fermium	101 (258) <b>Md</b> Mendelevium	102 (259) <b>No</b> Nobelium	103 (262) <b>Lr</b> Lawrencium



**VersaLab™** Temperature: 50 K–1000 K Field: ± 3 T  
**PPMS® DynaCool®** Temperature: 50 mK–1000 K Field: ± 9 T, ± 12 T, ± 14 T  
**FusionScope™** Correlated Microscopy Platform AFM, C-AFM and MFM  
**MPMS®3 EverCool®** Temperature: 0.5 K–1000 K Field: ± 7 T  
**OptiCool®** Temperature: 1.7 K–350 K Field: ± 7 T  
**PPMS®** Temperature: 50 mK–1000 K Field: ± 9 T, ± 14T, ± 16 T







